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Proceeding of The International Conference on Environmentally Friendly Civil Engineering Construction and Materials, 13-14 November 2014, Manado - Indonesia p. 416

1. Structure, geopolymer and other construction materials, 2. Water resources and environmental management, 3. Transportation and urban infrastructure, 4. Geotechnical engineering

ISBN: 978-602-71876-0-3

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Address: Jl. Kampus Barat – Bahu

Manado 95115

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TABLE OF CONTENTS

TITLE PAGE	I. seems
PREFACE	iii
LIST OF COMMITTEE	v
TABLE OF CONTENTS	vii
Keynote Speaches	
Geopolymer Concrete Application	
By B. Vijaya Rangan	I
Scientific Approach to Understand the Trans-boundary Management of Groundwater Resources in Kumamoto Area, Japan	9
By Jun Shimada	
Climate Change Impact on Urban Stormwater Runoff: Why We Need To Be Concerned By Ashantha Goonetilleke	19
Lime Based Materials in Construction: Experimental investigations for the development and validation of atomic models	20
By R. J. Ball, G. L. Pesce, S. C. Parker and R. J. Grant	49
Landslides: Anticipation and Prevention in Indonesia By Tommy Ilyas	37
Topic A Structure, Geopolymer and Other Construction Materials	
Decorative Pervious Concrete: A Solution for Aesthetically Sustainable Green	
Environment	41
By Tavio and Benny Kusuma	41
Parametric Study on the Compressive Strength of Palm Oil Fuel Ash (POFA) Geopolymer Mortar	
By Monita Olivia, Winda Astuti H Putri, Lita Darmayanti, Iskandar R Sitompul, Alfian Kamaldi and Zulfikar Djauhari	49
Effect of Nano Silica on Durability Properties of Concretes Containing Recycled Coarse	
Aggregates By Faiz Uddin Ahmed Shaikh and Heasam Odoh	57
Effect of Reactive Powder for Compressive Strength Concrete By Tavio and Candra Gunawan	65
Experimental Behaviour of RC Columns Externally Confined by Reinforcing Steel Bars Subjected to Axial compression	
By Utari Khatulistiani and Tavio	71
Recycling of Industrial By-products in the Construction Field in Oman - An Overview By Khalifa S. Al-Jabri	81

Timber Connections	
Finite Element Modeling and Experimental Tests of Parkia Speciosa Timber Connections By Yosafat Aji Pranata and Deny Anarista Sitorus	91
Performance Characteristics of Fiber-reinforced Concrete with Polyethylene Terephthalate (PET) Fiber; Part 1. Abrasion Resistance and Chloride Penetration By Ricky N Ratu and Nemkumar Banthia	
Compressive strength of Fly Ash Based Geopolymer Concrete By Riger Manuahe, Reky S. Windah and Dody M. J. Sumajouw	
Strength and Deformability of Confined Rectangular Reinforced Concrete Columns with Supplemental Pen-Binder By Anang Kristianto and Iswandi Imran	121
Effect of Nano-Silica Particle Addition on Compression Behaviour of Non-Stoichiometrically Cured Epoxy Resin By Markus Karamoy Umboh, Tadaharu Adachi, Masahiro Higuchi and Zoltan Major	
Statistical Analysis of Confined Columns with Varying Reinforcement Configuration and	
By Benny Kusuma, Tavio and Priyo Suprobo	141
AE-SiGMA Analysis in Split-Tensile Test of Fiber-Reinforced Concrete (FRC) at Meso- Scale By Mielke R.I.A.J Mondoringin1, Masayasu Ohtsu2	
Fly Ash and GGBS Based George	
By G.Mallikarjuna Rao, Venu M and T.D.Gunneswara Rao Influence of Sulfate Attack to Reactive Powder Concrete By Widodo Kushartomo, F.X. Supartono and Octavivia	
Preliminary Study of The Development of Volcanic Ash-Based Geopolymer Concrete The Effect of Superplant in the Pandaleke	
The Effect of Superplasticizer Variation toward the Slump of Fly Ash Based Geopolymer Concrete By Anggie Adityo Aer, Ronny E. Pandeleke and Dody M.J. Sumajouw Durability and Performance of Polymer Advanced Polymer Polym	
Durability and Performance of Polymer Modified Ferrocement in General Application Application of European Yield Model on Land	191
Application of European Yield Model on Joint Laterally Strength Analysis of Bamboo By Gusti Made Oka, Andreas Triwiyono, Suprapto Sigurous	
Strength Development of Concrete By Gidion Turu'allo and Marios N. Soutsos	
Improving the Mechanical Properties of Aluminum 6063 Structural Items by Varying the Chemical Composition By Indika De Silva, Asela Perera, Navinda Kodikara and Asanka De Silva.	. 223
A Coanka De Silva	233

	Study of Tensile Strength of Geopolymer Concrete with Amurang Fly Ash as a Source Material	
	By Andre K. Putra I. Fiha E.S. Paat, S.O. Dapas, R.S. Windah and S.E. Wallah	239
	A study of Green Construction Assessment of Two Buildings in Yogyakarta By Peter F Kaming	247
	Lightweight High-strength Concrete with Expanded Polystyrene Beads as Building Materials in Earthquake Zones	
	By Tengku Fitriani L. Subhan, Tommy A.M. Tilaar and Ravi Sri Ravindrarajah	257
	Development of Thermal Comfort Model in Minahasa Stilt House Industrial Production with Optimization the Effect of Outer Room Design By Debbie Augustien Jeannet Harimu	265
Ť	opic B	1140,710
	Vater Resources and Environmental Management	
	The Implication of Critical Shear Stress of Erosion on the Depth of Sediment in Brantas River – East Java	
	By Wati Asriningsih Pranoto and Abrar Riza	287
	Analysis of Treatment Performance of Constructed Stormwater Wetlands with Utilising a Simplified Conceptual Model	
	By Isri Ronald Mangangka, Ashantha Goonetilleke, Prasanna Egodawatta, Sukarno and Cin Jeane Supit	297
	Effect of Regularly Square Ribs Roughness on Turbulent Flow Structure in Cases Two and Three-dimensional Roughness in an Open Channel	
	By Sukarno, Lianny Hendratta, Isri Ronald Mangangka, Cindy Jeane Supit and Terunori Ohmoto	309
	Rheological Characteristics of Hyperconcentrated Sediment-laden Flow	
	By Liany A. Hendratta, Tiny Mananoma, Jeffry Sumarauw and Terunori Ohmoto	.317
	Identification of Water Resources at Botanical Garden in Pasir Impun, West Java, Using Georesistivity Method, Wenner Configuration	
	By Diyan Parwatiningtyas, Achmad Sjamsuri and Dasmo	323
	Water Resources Management by Modelling of Dam Operation By Cindy Jeane Supit, Isri Ronald Mangangka, Sukarno and Audy Supit	222
		355
	A Study on Performance of The Traditonal Protection to Reduce The Inundation Area in Saga Plain	
	By Ariestides K. T. Dundu, Koichiro Ohgushi and Hiroyuki Araki	341
T	opic C	
I	ransportation and Urban Infrastructure	
	Evaluation of Coastal Vulnerability Assessment and Coastal Management for Karangantu Coastal Area in Banten - Indonesia	
	By Olga Pattipawaej and Hanny Dani	351

Vard in Indonesia	
Arrangement of Environmentally Sound Ship Recycling Yard in Indonesia By Sunaryoand Dovan Pahalatua. Sound Ship Recycling Yard in Indonesia By Sunaryoand Dovan Pahalatua.	361
D. C. Strangement of Environmentary Source	
- 121 Open SKV POHCV TA	
Facility in a mational Alimona	
Study on Airline Hub Eastern Part of Indonesia Facing the ASEAN Open on Study on Airline Hub Eastern Part of Indonesia Facing the ASEAN Open on Study of Full Service Carrier Airlines in Sultan Hasanuddin International Airport) Case Study of Full Service Carrier Airlines in Sultan Hasanuddin International Airport) By Sakti Adji Adisasmita, Nur Ali, Mubassirang Pasra and Dantje Runtulalo Dender Areas Development (Case	2035
Case Study of Full Service Carrier Airlines in Suntain Page and Dantje Runtulaio	369
By Sakti Adii Adisasmita, Nur Ali, Mubassirang Pasia	
Enhancing Infrastructure in the Indonesia - Malaysia Border Areas Development (Case Studies in Entitions, Sangary District, West Kalimantan)	
Enhancing Infractructure in the Indonesia - Malaysia Bolder	
Studies in Entikong, Sanggau District, West Kalimantan)	256
By Tommer II. D. D. D. Etc. Curvani and Henni	
Studies in Entikong, Sanggau District, West Kalimantan) By Tommy Ilyas, Dwi Dinariana, Fitri Suryani and Henni	
1. The Case Study of Ividiado	
Toward Great Streets Using Urban Design Approach	385
Toward Great Streets Using Urban Design Approach: The Case By Febriane Paulina Makalew	-
Topic D	
Geotechnical Engineering	
- Control Engineering	
Powds of Load Test and FEM	
Load Settlement behavior of Bored Pile Based on The Result of Load Test and FEM	20.0
By Asriwiyanti Desiani and Angel Refanie	395
Recommendation of Viscosity Values for Mudflow	
By Budijanto Widjaja, David Wibisono Setiabudi and Ivan Arista	401
Jacob Widjaja, David Wibisono Setiabudi and Ivan Adisa	
Soil air Gas Survey at Malana VIII V. Landbary Gootharmal Field North	
Soil air Gas Survey at Makaroyen Village in Kotamobagu Geothermal Field, North	
Sulawesi, Indonesia	
By Hendra Riogilang, Oscar Hans Kaseke, Saartje Monintja and Cindy Jeane Supit	407
Bending Moment of Flexible Batter Piles in Sand under Horizontal Loads	
By Fabian J. Manoppo	1444
	413



INFLUENCE OF SULFATE ATTACK TO REACTIVE POWDER CONCRETE

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ABSTRACT

Sulfates are often present in groundwaters, particularly when high proportions of clay are present in the soil, and sea water has sulfates as a major constituent. Sulfate attack on concrete is actually a rather complex process which may involve a number of secondary processes. This is accompained by a very large increase in solid volume, which cause a volume expansion within the paste and which generates accompanying internal stress and ultimately leads to cracking

The use of special concrete such as reactive powder concrete can improve sulfate resistance. Ordinary portland cemen (OPC) or portland compocite cement (PCC) can be used instead of a type II cement, type II of cement that have the best performace from sulfate attack on reactive powder

concrete.

KEYWORDS: concrete, sulfate, cement, resistance, cracking.

1. INTODUCTION

The most widespread and common form of chemical attack is the action of sulfates on concrete. Groundwaters may have local concentrations of sulfates in the vicinity of industrial wastes such as mine tailings, slag heaps, and rubble fills. Sulfates present in rainwater from air pollution, or produced by growths, may coause slow deterioration even in concrete above ground (Mindess and Young, 1981).

External sulfate attack is the more common type and typically ocurs where water containing dissolved sulfate penetrates the concrete. A fairly well-defined reaction front can often be seen in polished sections; ahead of the front the concrete is normal, or near-normal. Behind the reaction front, the composition and microstructure of the concrete will have changed these changes may vary in type or severity but commonly include extensive cracking, expansion, loss of bond between the cement paste and agregate. The effect of these changers is an overall loss of concrete strength (Winter, 2012).

Reactive Powder Concrete (RPC) are well-known for their high strength resistance. Their very low porosity gives them important durability and transport properties and makes them potentially a suitable material for the agressive inveroment such as ground water and seawater (Matte and Moranville, 1999).

This paper describes a study which was carried out to better understand the behaviour of the RPC especialy compressive strength and flexture strength to the influence of sulfate attack.

2. REACTIVE POWDER CONCRETE

RPC composed by cement, sand, quartz powder, silica fume, steel fiber (optional) and superplasticizer. The superplasticizer, used at its optimal dosage, decreases the water to cement ratio (w/c) while improving the workability of the concrete. Convensional concrete aggregates form a set of rigid inclusions. On application of a compression force shear and tensile stresses appearing at the paste/aggregate interface generate cracks in paste. The size of these cracks is

related to extent of the zone under tensile or shear stress. For RPC, with eliminated coarse aggregat a major reduction is obtained in the size of microcraks of the following origins: external load, autogenous shringkage and differential expansion between paste and aggregate under effect of heat treatment (Richard and Cheyrezy, 1995).

Elimination of coarse aggregates combined to optimization of granular mixture allows the obtention of an homogeneous and dense cementitious matrix that exhibits high mechanical

performances (Kushartomo et al, 2013).

3. SULFATE ATTACK

Sulfate attack on concrete is actually a rather complex process which may involve a number of secondery processes, however, laboratory and field experiences have definitely established a correlation between the C3A content of a portland cement and its susceptibility to sulfate attack. The major cause is soulfoaluminate corrosion, in withe ettringite is formed from monosulfoaluminate (Mindes and Young, 1981):

$$C_4A\overline{S}H_{12} + 2C\overline{S}H_2 + 16H$$
 $C_6A\overline{S}H_{32}$

This is accompained by a very large increase in solid volume, which cause a volume expansion within the paste and which generates accompanying internal stress and ultimately leads to cracking.

Mindess and Young (1981) describes the sulfate attack into 3 processes:

Penetration sulfate ion
 The first process is the diffusion sulfate ions into the pores of the concrete, whice is controlled by the permeability coefficient and the diffusion coefficient of the sulfate ions.

2. Gypsum corrosion In its initial stages, gypsum corrosion may actually be beneficial, since gypsum is more soluble than calcium hydroxide and the dissolution-crystallization reaction will allow gypsum to crystallize without expansion. Equation of gypsum corrosion is as follows:

$$CH + SO_4^{2-} + 2H_2O \rightarrow C\overline{S}H_2 + 2OH^{-}$$

Calcium hydroxide + Sulfate + Water → Gypsum + Hydroxide

3. Sulfoaluminate corrosion

As sulfoaluminate corrosion causes internal cracking, the diffusion coefficient of the concrete will be increased, thereby accelerating further sulfate attack. Ettringite in this case is formed from monosulfoaluminate can be seen below.

$$C_4A\overline{S}H_{12}$$
 + $2C\overline{S}H_2 + 16H \rightarrow C_6A\overline{S}_3H_{32}$

Calcium Aluminate Hydrate + Gypsum + Water → Ettringite

4. RESEARCH METHODE

The RPC used in this study was prepared from the following components, quartz sand with maximum size 0,6 mm was used as a fine aggregate. The specific gravity of quartz sand is 2,65 g/cm³ polycarboxylate based superplasticizer (SP) is conformity with ASTM C 494-81, type F was used in this study. The physical, chemical and mechanical properties of cement (type 1, type II and PCC), properties of silica fume and quartz powder are presented in Table 1. Furthermore, to accelerated chemical reaction the highest magnisium sulfat (MgSO4) concentration was used. Various mixture designs (mass ratio with respect to cement) silica fume, quartz as fine aggregate, quartz powder and superplasticizer are presented in Table 2.

Table 1. Physical and mechanical properties of cement, silica fume, and quartz powder

	Type I	Type II	PCC	Silica Fume	Quartz powder
Chemical composition					
SiO ₂	20.15	22,6	22,5	98,0	98,6
Al ₂ O ₃	5,65	3,4	6,0	0,29	0,7
Fe ₂ O ₃	2,15	4.2	3,5	0,03	0,03
CaO	62,92	64,9	60,0	0,2	
MgO	1,14	0,90	2,25	0,09	
$Na_2O + K_2O$	1,15	0,05	0,45	0,23	
SO ₃	2,52	2,2	2,25	0,08	
Cl	0,001	0,01	0,015	0,01	
F.CaO (%)	0,52	1,0	1,0		
Physical propertie	s				
Specific gravity	3,13	3,15	3,05	2,20	2,65
Initial setting (min)	130	130	130		
Final seting (min)	210	280	305		
Surface area (cm²/g)	380	380	385		
Compressive strength	(MPa)				
3 days	29,2		23,0		
7 days	43,2		30,0		
28 days	51,9		41,0		

Table 2. Composition of the mixture investigation

Materials	Type I	Туре П	PCC	Type I	Туре ІІ	PCC
Cemen (kg)	757,8	759,0	753,0	730,1	731,2	725,7
Water (kg)	151,6	151,8	150,6	182,5	182,8	181,4
Silica fume (kg)	189,5	189,7	188,3	182,5	182,8	181,4
Quartz powder (kg)	189,5	189,7	188,3	182,5	182,8	181,4
Fine aggregate (kg)	1136,7	1138,5	1129,5	1095,2	1096,8	1088,6
Superplasticizer (kg)	22,7	22,8	22,6	21,9	21,9	21,8
Water/cement	0,20	0,20	0,20	0,25	0,25	0,25
Silica fume/cement	0,25	0,25	0,25	0,25	0,25	0,25
Quartz powder/cement	0,25	0,25	0,25	0,25	0,25	0,25
Fine aggregate/cement	1,50	1,50	1,50	1,50	1,50	1,50
Superplasticizer/cement	0,03	0,03	0,03	0,03	0,03	0,03

The concrete was mixed, cast, and vibrated like conventional concrete. Dry concrete powder components are mixed for 3 to 5 minutes, water is added with half on the volume superplasticozer and mixed for 3 minutes, and the second half of the volume of the superplasticizer is added and mixed for 5 minutes. The cylindrical samples were 750 mm in diameter and 150 mm long. The following curing treatment was applied to stabilize the mechanical and physical properties of speciment in steam at 90°C for 12 hours. The Sulfate attack test simulates the worst attack on the cylindrical samples used saturated solution of MgSO4. In this severe test, the samples

are immersed in sulfate solution bath maintained 24 hours, washing with hot water, 24 hours drying oven at 105 °C, immersed again continously for 7 times. Specimen will be tested when it's already 28-days.

5. RESULT AND ANALYSIS

These chemical reactions can lead to expansion and cracking of concrete, and/or the loss of strength and elastic properties of concrete is given in Fig. 1 dan Fig. 2, its represents compressive strength of RPC containing different treatment. The form and extent of damage to concrete will depend on the sulfate concentration, the type of cations (eg sodium or magnesium) in the sulfate solution, the pH of the solution and the microstructure of the hardened cement matrix (Sarkar et al, 2010).

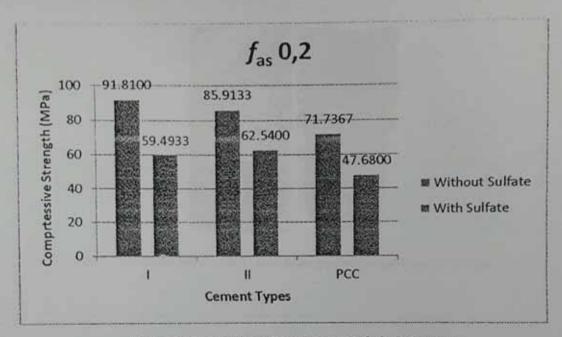


Figure 1. Compressive Strength with fat 0,2 diagram

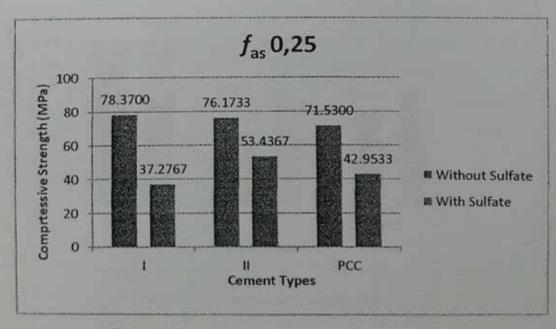


Figure 2. Compressive Strength with far 0,25 diagram

Figure 3 represent mass reduction indicated of sulfate attack. It can be seen face differences of RPC samples. There are no holes are formed for normal sample and any holes for sample immersed in sulfate solution, it caused reduces mass of RPC sample as given on Fig.4. Holes due to the formation of ettringite compounds that are expanding. Some cements are more susceptible to magnesium sulfate than sodium sulfate, the key mechanism is the replacement of calcium in calcium silicate hydrates that form much of the cement matrix. This leads to a loss of the binding properties. Formation of brucite (Mg(OH)₂) and magnesium silicate hydrates is an indication of such attack.



Figure 3. Specimen before (left) and after (right) immersion in sulfate

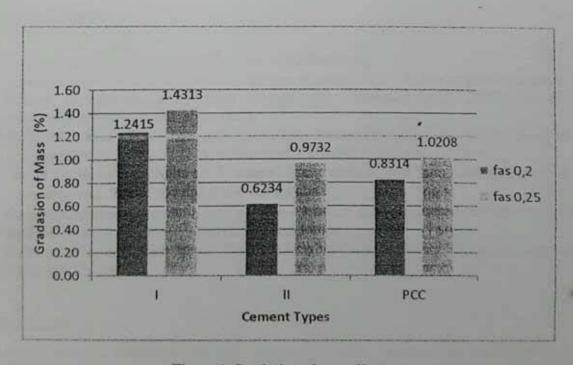


Figure 4. Gradation of mass diagram

In the composition of the cement, C₃A compound is a compound that was attacked by sulfate Composition of C₃A in type II cement is the least as much as 6%, so the cement Type II

said to be good for sulfate attack. If C₃A met with gypsum, it form ettringite that cause the crack expands and causes the compressive strength of concrete decreased.

It can be seen from the results of compressive strength test, specimens with Type II cement mixture produces % reduction in compressive strength at least compared to cement Type I and PCC. So it can be said that the Type II cement is the cement that has good resistance to sulfate compared with Type I cement and cement PCC.

C₃A in the cement compounds also affect the sulfate reaction which affected by gypsum. The main initial reaction of C₃A and gypsum reacts with water to form trisulfoaluminat / ettringite. If the sulfate is all consumed before C₃A completely hydrate, then ettringite transform to another calcium sulfoaluminate hydrate containing less sulfate called monosulfoaluminate.

$$C_3A + 3C\overline{S}H_2 + 26H \rightarrow C_3A.3C\overline{S}_3.32H$$

 $C_3A + 3C\overline{S}H_2 + 26H \rightarrow C_6A\overline{S}_3H_{32}$
 $2C_3A + C_6A\overline{S}_3H_{32} + 4H \rightarrow 3C_4A\overline{S}H_{12}$

At higher concentrations of sulfate, gypsum corrosion can contribute greater. Magnesium Sulfate can be even more aggressive because of the possibility of additional corrosive reaction due to the presence of magnesium ions, which can decompose both C-S-H and the calcium sulfoaluminates (Mindes and Young, 1981):

$$C_3S_2H_3 + 3M\overline{S} \rightarrow 3C\overline{S}H_2 + 3MH + 2SH_x$$

 $C_4A\overline{S}H_{12} + 3M\overline{S} \rightarrow 4C\overline{S}H_2 + 3MH + AH_3$

6. CONCLUSION

Based on the research, we can conclude some of the following:

- Reactive Powder Concrete with a mixture of cement type II has a good resistance against sulfate, because the compressive strength only decreased by 27.2057% on the highly concentrated (saturated) sulfate attack.
- Reactive Powder Concrete is good for resistance to sulfate attack due to the weight loss of the test specimen is not up 1.5% to sulfate attack that has a high concentration.
- Based on this research, Type II cement is the cement that has good resistance to sulfate compared with Type I cement and cement PCC.
- Reactive Powder Concrete is best used on structures that are directly related to ground water and sea water, for examples foundation and harbor.
- In responding to sulfate attack, Reactive Powder Concrete with PCC cement more efficient than type II cement.

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CERTIFICATE OF ATTENDANCE

AWARDED TO

DR. WIDODO KUSHARTOMO

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